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own symbol, which is a capital letter M. A solution that has the concentration 5 mol/L would be called a 5 M solution or said to have a concentration value of 5 molar.

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Molarity By Dilution Chemistry If8766 where the subscripts “1” and “2” refer to the solution before and after the dilution, respectively. Since the dilution process does not change the amount of solute in

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the solution,  $n_1 = n_2$ . Thus, these two equations may be set equal to one another:  
 $[M_1L_1 = M_2L_2]$  This relation is commonly referred to as the dilution equation.

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