

Gas Law Study Guide Answers

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The pressure of a gas is directly proportional to the temperature if the volume remains constant What is the formula for the combined gas law? $P_1V_1/T_1=P_2V_2/T_2$

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26. Ideal Gas Law and Stoichiometry Use the following reaction to answer the next few questions: $2 \text{C}_8\text{H}_{18}(\text{l}) + 25 \text{O}_2(\text{g}) \rightarrow 16 \text{CO}_2(\text{g}) + 18 \text{H}_2\text{O}(\text{g})$ The above reaction is the reaction between gasoline (octane) and oxygen that occurs inside automobile engines. 29) If 4.00 moles of gasoline are burned, what.

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The volume of a gas increases as the pressure on that gas decreases. When a system that's in equilibrium is disturbed, the system adjusts itself to reduce the change. The effusion rate of a gas is...

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Gas Laws Answer Study Guide Chemistry

The average kinetic energy of a gas is expressed by the formula: $KE_{ave} = 3RT/2$ where KE_{ave} = average kinetic energy R = ideal gas constant T = absolute temperature The average velocity or root mean square velocity of individual gas particles can be found using the formula $v_{rms} = [3RT/M]^{1/2}$ where v_{rms} = average or root mean square velocity R = ideal gas constant T = absolute temperature M = molar mass

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Gas Laws Study Guide Answer A gas is a state of matter with no defined shape or volume. Gases have their own unique behavior depending on a variety of variables, such as temperature, pressure, and volume. While each gas is different, all gases act in a similar matter. This study guide

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