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Q. The Quantum Mechanical Model of the Atom describes the electron's probable location around the

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nucleus in a 3-D cloud called a(n) \_\_\_\_.

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Electrons are found in certain orbits located around the nucleus. Every electron has a fixed energy in certain energy levels: they are said to be



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quantized. Electrons farther away from the nucleus have higher energy than electrons closer to the nucleus. Energy levels are closer together the farther away from the nucleus.

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Electrons are added as well, but ...

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The orbital diagram for a ground-state nitrogen atom  
is  $1s^2 2s^2 2p^3$ . A: 5. The number of orbitals in a d subshell  
is A: 5 6. The maximum number of electrons that can

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occupy an energy level described by the principal quantum number,  $n$ , is  $A: 2n^2 - 7$ . A ground-state atom of manganese has \_\_\_ unpaired electrons and is \_\_\_\_\_. A: 5, paramagnetic 8.

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electrons exhibit properties of both particles and

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waves. d. the chemical properties of elements can be grouped according to periodicity but physical properties cannot. \_\_\_\_ 28. Elements in a group or column in the periodic table can be expected to have similar. ... Chemistry Chapter 5 Exam ...

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