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absolute temperature  
from 298 K. Calculate

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The activation energy for

the reaction,  $2 \text{HI}(\text{g})$

$\longrightarrow \text{H}_2 + \text{I}_2(\text{g})$  is

$209.5 \text{ kJ mol}^{-1}$  at

581 K. Calculate the

fraction of molecules

of reactants having

energy equal to or

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increasingly more difficult challenges in reactor design. The first one-third of the text emphasizes general principles of chemical reaction kinetics, setting the stage for the subsequent treatment of reactors intended to carry out homogeneous reactions,

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